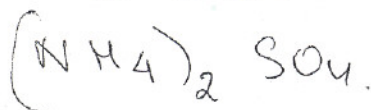
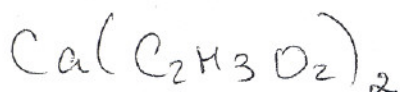
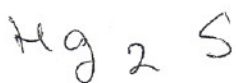
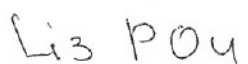


Additional Exercises

(Naming)

Give names to the following compounds:



Write formulas:

aluminum hydroxide

mercury(I) bromide

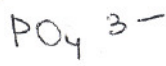
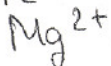
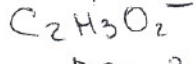
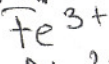
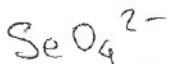
sodium hypobromite

calcium hydride

hydrobromic acid

tetraphosphorus
hexasulfide

Form neutral molecules from these ions



Key

iron(II) hydroxide

barium perchlorate

lithium phosphate

mercury(I) sulfide

calcium acetate

potassium chromate

ammonium sulfate

$\text{Al}(\text{OH})_3$

Hg_2Br

NaBrO

CaH_2

HBr

P_4S_6

$\text{Al}_2(\text{SeO}_4)_3$

$\text{Fe}(\text{C}_2\text{H}_3\text{O}_2)_3$

$\text{Mg}_3(\text{PO}_4)_2$